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MASS SPECTROMETRY OF π-COMPLEXES OF TRANSITION METALS

VIII. CrS⁺ IONS IN THE MASS SPECTRUM OF π -THIOPHENECHROMIUM TRICARBONYL

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Summary

The fragmentation of π -thiophenechromium tricarbonyl under electron impact has been examined. The formation of the rearrangement ion CrS⁺ has been observed.

Fragmentation of thiophenechromium tricarbonyl (I) under electron impact is similar to the degradation of benzenechromium tricarbonyl (II) [1,2] in that successive elimination of three carbonyl groups and of the ligand molecule occurs and results in the ions $C_4H_4SCr(CO)_n^+$ (n is 0 to 2) and Cr^+ , respectively (Table 1, Scheme 1). It should be noted that mass spectra of thiophene and benzenechromium tricarbonyls are very similar (Table 1).

SCHEME 1

$$C_4H_4SCr(CO)_3^+\xrightarrow{-CO} LCr(CO)_2^+\xrightarrow{-CO} LCr(CO)^+\xrightarrow{-CO} LG^+ \xrightarrow{*} Cr^+$$

$$C_4H_4S^+ \xrightarrow{*} CrS^+$$

TABLE 1

MASS SPECTRA OF THIOPHENECHROMIUM TRICARBONYL (I) AND BENZENECHROMIUM

TRICARBONYL (II) (70 eV)

	Ion	P ⁺	(P — CO)+	(P - 2CO)+	LCr ⁺	L+	(CrS) ⁺	Cr ⁺
ı	m/e % Σ	220 0.093	192 0.021	164 0.028	136 0.138	84 0,050	84 0.025	52 0.438
II	m/e % Σ	214 0.120	186 0.007	158 0.014	130 0.180	78 0.066		52 0.510

TABLE 2

EXACT MASSES AND THE C₁S⁺/C₄H₄S⁺ RATIOS IN THE MASS SPECTRUM OF TIOPHENECHROMIUM TRICARBONYL

	Masses Found (calcd.)	Δ	CrS ⁺ /C ₄ 70 eV	H ₄ S ⁺ at 15 eV		
C ₄ H ₄ S ⁺	84.0083 (84.0034)	0.0044	0.5	1	T _a i	
CrS+	83.9167 (83.9126)	0.0041		-		•
SCr(CO) ⁺	111.9028 (111.9075)	0.0047		-		

$$\begin{bmatrix} Cr(CO)_n \end{bmatrix}^{+} \qquad \begin{bmatrix} S \rightarrow Cr(CO)_n \end{bmatrix}^{+} \qquad \begin{bmatrix} Cr(CO)_n \end{bmatrix}^{+} \qquad \begin{bmatrix} S \rightarrow Cr(CO)_n \end{bmatrix}^{+} \qquad \begin{bmatrix} S \rightarrow$$

The mass spectrum of thiophenechromium tricarbonyl displays intense ions at m/e of 84 assignable to the ligand ion C_4H_4S . Measuring the exact masses has shown, however, that the peak is a doublet consisting of $C_4H_4S^+$ and CrS^+ , the CrS^+ contribution in the m/e 84 peak increasing with a decrease in the ionising electron energy (Table 2). The formation of CrS^+ ions suggests that, along with the expected π -coordination of $C_4H_4CrS^+$ (III), at least part of the $C_4H_4SCr^+$ is subject to σ -type coordination (IV), or that both π - and σ -coordination (V) operate in the ion $C_4H_4SCr^+$.

Ions containing a Cr—S bond may arise not only following the complete decarbonylation of the parent ion but also at intermediate steps of the elimination. This agrees with the $SCr(CO)^+$ ion found in the spectrum. The $SCr(CO)^+$ ion $(n ext{ is } 2 ext{ or } 3)$, if they exist at all, are of very low intensity.

Experimental

Mass spectra were recorded on an AEI MS-30/DS-50 mass spectrometer at 20°C. The ionising chamber temperature was 100°C and the ionising energy was 70 or 15 eV.

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